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Aminohydroxyacetone synthons: versatile intermediates for the organocatalytic asymmetric aldol reaction†‡

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A practical method for the synthesis of 1,3-aminohydroxyacetone synthons was developed, and their utility in the organocatalytic asymmetric aldol reaction was demonstrated in a short synthesis of azasugars.

Introduction

Protected 1,3-dihydroxyacetone compounds are versatile intermediates for the synthesis of biologically interesting natural products with polyhydroxylated functionality.¹ The key structural unit of 1,3-amino alcohol is very relevant to this chemistry, and has been found in many antibiotics and other biologically active natural products.² In addition, 1,3-amino alcohols and their derivatives can play an important role in asymmetric synthesis as chiral ligands, as chiral auxiliaries, as resolving agents, and as phase-transfer catalysts.³ Hence, there are considerable synthetic challenges to constructing these frameworks by the use of several strategies, but most of these are based on diastereoselective or enantioselective reduction of β-amino ketones or β-hydroxy imines/oximes.⁴ There are therefore still some limitations in the direct and efficient methods for the construction of these valuable species. **Commission Commission**
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 Aminohydroxyacetone synthens: versatile intermediates for the

organocatalytic asymmetric aldol reaction⁺#

Organocatalytic asymmetric transformations have recently attracted a great deal of attention from synthetic chemists as powerful and fascinating tools because of their mildness, high efficiencies, and environmentally friendly characteristics.⁵ In our extensive efforts in this field, 6 we were particularly interested in the development of new designer aminohydroxyacetone synthons as analogs of 1,3-dihydroxyacetone derivatives. To our surprise, however, a search of the literature revealed that there is no precedent regarding the preparation of these molecules.

In this paper, we describe a new practical method for the synthesis of 1,3-aminohydroxyacetone synthons and their use in organocatalytic asymmetric aldol reactions.

Results and discussion

Our synthetic strategy is outlined in Scheme 1.

Scheme 1 Synthesis of protected aminohydroxyacetones.

First, we started with the known allyl bromide 1 , DIBAL-H reduction of the ester followed by exposure to an excess of aqueous ammonia and subsequent protection of the primary amine as tert-butyl carbamate or methyl carbamate gave 2a or 2b in good yield. Next, treatment of this substrate with 2-methoxypropene in the presence of a catalytic amount of PPTS smoothly gave the corresponding acetonide 3. Finally, conversion to the target molecule 4 was achieved via oxidative cleavage of the double bond under either ozonolysis (Method A for 4a) or Lemieux–Johnson conditions (Method B for 4b). This overall process is easy for a multi-gram-scale preparation and for the attachment of other nitrogen protective groups such as benzyl carbamate.

Next, we examined the aldol reaction of 4 with 0.5 equiv of p nitrobenzaldehyde using L-proline (30 mol%) in dimethyl sulfoxide (DMSO) at room temperature. Unfortunately, however, no

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Table 1 Organocatalytic asymmetric aldol reaction of 4: optimization study

 a Conditions: 4 (1 mmol), p-nitrobenzaldehyde (0.5 mmol) in the solvent (2.5 mL) in the presence of 30 mol% of the catalyst. \overline{b} Isolated yield. c Determined by ¹H NMR. d Determined by chiral HPLC analysis using a Chiralcel OD-H after converting to the corresponding acetate. ^e 100 mol% of the catalyst was used. $f_{1,3}$ -Oxazolidine **C** was isolated in 40% yield. $\frac{g}{2}$ 500 mol% of H₂O was added as an additive.

reaction was observed even after 1 week (Table 1, entries 1 and 3). The use of 100 mol% of L-proline caused the parasitic reaction of L-proline^8 and gave bicyclic 1,3-oxazolidine C in 40% yield (based on L-proline) (Table 1, entry 2). Instead, the use of L-proline-derived tetrazole \mathbf{B}^9 as a catalyst was found to be favorable for our purpose. Thus, the reaction of 4a with 0.5 equiv of p -nitrobenzaldehyde using 30 mol% of catalyst **B** in *i*-PrOH at room temperature gave the adduct 5a in 81% yield with excellent diastereoselectivity $(syn : anti = 2 : 98)$ and high enantioselectivity (90% ee), but the reaction required 5 days to reach completion (Table 1, entry 4). Interestingly, the reaction was rather sensitive to the solvent used, and the reaction rates were relatively slow in DMSO, DMF, and MeCN (Table 1, entries 5–7, 10). After screening, we found that the reaction with 4b in i-PrOH was the best in terms of reaction time, product yield, and diastereo- and enantioselectivity (Table 1, entry 11). In contrast to our previous observations regarding proline-catalysis, 10 the addition of water significantly decreased the enantioselectivity (Table 1, entry 12).

At this stage, the regiochemistry of the reaction course and the absolute configuration of the major diastereomer of 5 were both unknown. To solve these problems, 5b was reduced with NaBH₄ in THF–MeOH followed by esterification using 2.5 equiv of

Scheme 2 Derivatization of 5b ($P = CO₂Me$).

Fig. 1 Structure of the dibenzoate stereoisomer 7 by X-ray analysis.

 p -bromobenzoyl chloride– i -Pr₂NEt in CH₂Cl₂ (Scheme 2). Fortunately, one of the dibenzoate stereoisomers, 7, was obtained as a crystalline substance (mp $170-172$ °C) and its structure was unambiguously determined by X-ray crystallographic analysis (Fig. 1). 11

As evident from this result, carbon–carbon bond-formation was highly regioselective, and the reaction took place only at the oxygen-containing α-side of aminohydroxyacetone 4. On the other hand, the absolute stereochemical sequence exactly dictates well-established enamine-based asymmetric organocatalysis.¹² Consequently, in the present system, enamine formation via deprotonation at the α-carbon might be considered to be facilitated by the enhanced acidity due to the presence of an electronegative oxygen atom.

With the optimal conditions in hand, we then examined a variety of aromatic and heteroaromatic aldehydes to establish the general utility of this synthetic methodology. All reactions were performed in i-PrOH at room temperature in the presence of 30 mol% of catalyst **B** (Table 2).^{13,14}

Table 2 Organocatalytic asymmetric aldol reaction of 4b: generality⁴

ArCHO	catalyst B (30 mol\%) i-PrOH, rt	
		MeO ₂ C ²

^a Conditions: **4b** (1 mmol), ArCHO (0.5 mmol) in *i*-PrOH (2.5 mL) in the presence of 30 mol% of the catalyst. b Isolated yield. The absolute</sup> configuration of the products was surmised by analogy. "Determined by ${}^{1}H$ NMR. d Determined by chiral HPLC analysis using a Chiralcel OD-H after converting to the corresponding acetate. ϵ MeCN was used as the solvent.

Various aldehydes reacted smoothly with 4b in moderate yields (up to 93%) with excellent diastereo- (up to $syn : anti = 2 : 98$) and enantioselectivity (up to 97%) (Table 2, entries 1–8). This method is particularly useful for reactive aldehydes bearing electron-withdrawing groups, with a good level of chemical yield (Table 2, compare entries 1–4 with entries 5–8).

To demonstrate the synthetic value of this organocatalytic asymmetric protocol, polyhydroxylated pyrrolidine alkaloids were targeted, since these molecules have attracted considerable attention as mimics of sugars and as potential pharmaceutical drugs.¹⁵ Our four-step sequence is outlined in Scheme 3.

Mesylation of $5b$ by treatment with MsCl–Et₃N followed by reduction with NaBH4 in THF–MeOH provided monoalcohol 16 in good yield with high stereoselectivity (85 : 15). After deprotection of an acetonide function, intramolecular cyclization was affected by exposure to 1,8-diazabicyclo[5.4.0]undec-7-ene

Scheme 3 Short synthesis of aza-sugar 18 ($P = CO₂Me$).

(DBU) in THF at room temperature to afford the desired pyrrolidine derivative 18 with an almost complete retention of stereochemistry (60% total yield). The relative configuration of this product was assigned by NOESY experiments, suggesting at the final stage in this reaction sequence the S_N 1-type pyrrolidinering cyclization at the benzylic position.

Finally, we briefly examined the applicability of 4 to the threecomponent Mannich-type reaction using p-nitrobenzaldehyde and p-anisidine (Table 3). In this case, the use of 4a as a donor in DMF was most favorable, and the expected 1,4-diamine derivative 19 was obtained in 63% yield with high diastereo- $(syn : anti = 84 : 16)$ and enantioselectivity $(88\% \text{ ee})$ (Table 3, entry 2).^{13,16}

Table 3 Organocatalytic asymmetric synthesis of Mannich adduct 19^a

4a	CHO $\ddot{}$ NO ₂	NH ₂ OMe	catalyst B (30 mol%) solvent, rt Boc ^N	PMP HN 19	NO,
Entry	Solvent	Time (h)	Yield ^b $(\%)$	syn : anti ^c	ee^{d} (%)
1	DMSO	72	71	68:32	68
$\overline{2}$	DMF	96	63	84:16	88
3	NMP	96	55	75:25	88
$\overline{4}$	THF	96	23	81:19	ND^e

 a Conditions: **4a** (0.75 mmol), p-nitrobenzaldehyde (0.5 mmol), panisidine (0.5 mmol) in the solvent (1.7 mL) in the presence of 30 mol[%] of the catalyst. ^b Isolated yield. ^c Determined by ¹H NMR. $\frac{d}{d}$ Determined by chiral HPLC analysis using a Chiralpak AD after deprotection of an acetonide function followed by mono-acetylation. ^e Not determined.

Conclusions

In conclusion, we have developed a new practical method for the synthesis of aminohydroxyacetone synthons, and demonstrated their versatile utility in organocatalytic asymmetric aldol and Mannich reactions. In addition, this methodology was successfully applied to the short synthesis of aza-sugars in optically active forms. We believe that this method should provide an expeditious route to a significant class of 1,3-amino alcohol derivatives. Further studies to extend the scope of this method are now in progress.

Experimental section

2-(Bromomethyl)prop-2-en-1-ol (1′)

To a solution of 1 (5.23 g, 29.4 mmol) in CH_2Cl_2 (200 mL) was added at −78 °C DIBAL-H (1.03 M in hexane; 60 mL, 61.8 mmol) over 0.5 h. After stirring for 1 h, the mixture was allowed to warm to 0 °C and quenched by addition of MeOH. Subsequent addition of saturated Rochelle's solution produced a gelatin-like solid, which was stirred at room temperature until the slurry was re-dissolved and a separation of the layers was observed. The aqueous layer was separated and extracted thoroughly with CH_2Cl_2 . The combined extracts were washed with saturated NaCl, dried (Na_2SO_4) , and concentrated under reduced pressure. The residue was purified by silica-gel column chromatography (eluted with hexane–Et₂O = 2 : 1) to afford the bromoalcohol 1′ (3.5 g, 79%) as a colorless oil; FTIR (KBr) ν 3366, 1210, 1059, 1027, 920 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.96 (1H, br), 4.05 (2H, s), 4.29 (2H, s), 5.26 (1H, m), 5.31 (1H, m); ¹³C NMR (100 MHz, CDCl₃) δ 32.8, 63.2, 115.6, 144.6; Anal. Calcd for C4H9BrO: C, 31.82; H, 4.67. Found: C, 32.20; H, 4.43.

tert-Butyl 2-(hydroxymethyl)allylcarbamate (2a)

To a solution of 28% aq NH₃ (60 mL) was added dropwise a solution of $1'$ (2.8 g, 18.7 mmol) in MeOH (9 mL) at 0 °C over 10 min. After stirring for 1 h, the mixture was concentrated under reduced pressure. The residue was re-dissolved in a mixture of 1,4-dioxane–H₂O (40 mL, $3:1$), and Boc₂O (4.5 g, 20.5 mmol) and Et_3N (5.2 mL, 37.3 mmol) were added at room temperature. After stirring for 1 h, $H₂O$ was added and the mixture was extracted with AcOEt. The combined extracts were dried (Na2SO4) and concentrated under reduced pressure. The residue was purified by silica-gel column chromatography (eluted with hexane–AcOEt = $2:1$) to afford 2a (2.87 g, 82%) as a colorless oil; FTIR (KBr) v 3352, 1694, 1524 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.45 (9H, s), 3.02 (1H, br), 3.80 $(2H, d, J = 6.1 \text{ Hz})$, 4.10 $(2H, d, J = 2.4 \text{ Hz})$, 4.87 $(1H, br), 4.99$ (1H, s), 5.06 (1H, s); ¹³C NMR (100 MHz, CDCl₃) δ 28.4 (\times 3), 42.8, 64.0, 79.9, 112.2, 146.2, 156.8; Anal. Calcd for C9H17NO3: C, 57.73; H, 9.15; N, 7.48. Found: C, 58.08; H, 9.53; N, 7.58.

Methyl 2-(hydroxymethyl)allylcarbamate (2b)

To a solution of 28% aqueous $NH₃$ (60 mL) was added dropwise a solution of 1′ (3.1 g, 20.7 mmol) in MeOH (9 mL) at 0 °C

over 10 min. After stirring for 1 h, the mixture was concentrated under reduced pressure. The residue was re-dissolved in a mixture of 1,4-dioxane–H₂O (40 mL, 3:1), and ClCO₂Me $(1.91 \text{ mL}, 22.7 \text{ mmol})$ and Et₃N $(5.75 \text{ mL}, 41.3 \text{ mmol})$ were added at room temperature. After stirring for 1 h, the solvent was removed under reduced pressure and the residue was extracted with AcOEt. The combined extracts were dried (Na_2SO_4) and concentrated under reduced pressure. The residue was purified by silica-gel column chromatography (eluted with hexane– AcOEt = 1:1) to afford 2b (2.11 g, 72%) as a colorless oil; FTIR (KBr) v 3337, 1706, 1541, 1265 cm⁻¹; ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$ δ 3.19 (1H, br), 3.69 (3H, s), 3.85 (2H, d, J $= 6.1$ Hz), 4.11 (2H, d, $J = 3.9$ Hz), 5.02 (1H, s), 5.10 (1H, s), 5.26 (1H, br); ¹³C NMR (100 MHz, CDCl₃) δ 43.1, 52.3, 63.9, 112.2, 145.6, 157.7; Anal. Calcd for $C_6H_{11}NO_3$: C, 49.65; H, 7.64; N, 9.65. Found: C, 49.49; H, 8.02; N, 9.75. nor Unive. After timing for 1), then nixture was concentrated for S k, the mixture was concertrated university on the mixture was reposition of the mixture was concentrated by the mixture of 1.4 Accounts (1.4 March 2012 O

tert-Butyl 2,2-dimethyl-5-methylen-1,3-oxazinane-3-carboxylate (3a)

To a solution of 2a (2.01 g, 11.3 mmol) in benzene (56 mL) were added 2-methoxypropene (4.2 mL, 45.2 mmol) and PPTS (284 mg, 1.13 mmol) at room temperature, and the mixture was stirred for 9 h. After quenching by addition of 1 N NaOH, the mixture was extracted with AcOEt. The combined extracts were washed with saturated NaCl, dried (Na₂SO₄), and concentrated under reduced pressure to afford 3a (2.46 g, 96%) as a colorless oil, which was used for the next reaction without further purification; FTIR (KBr) *v* 1696, 1384, 1366, 1245, 1164, 1065 cm⁻¹;
¹H NMP (400 MHz, CDCL) § 1.47 (9H s), 1.61 (6H s), 4.09 ¹H NMR (400 MHz, CDCl₃) δ 1.47 (9H, s), 1.61 (6H, s), 4.09 (2H, s), 4.35 (2H, m), 4.77 (1H, m), 4.93 (1H, m); 13C NMR (100 MHz, CDCl₃) δ 25.0 (×2), 28.4 (×3), 44.7, 63.0, 79.9, 88.1, 106.4, 142.7, 153.5; Anal. Calcd for $C_{12}H_{21}NO_3$: C, 63.41; H, 9.31; N, 6.16. Found: C, 63.80; H, 9.54; N, 6.17.

Methyl 2,2-dimethyl-5-methylen-1,3-oxazinane-3-carboxylate (3b)

To a solution of 2b (3.0 g, 21 mmol) in benzene (100 mL) were added 2-methoxypropene (7.7 mL, 83 mmol) and PPTS (1.04 g, 4.1 mmol) at room temperature, and the mixture was stirred for 5 h. After quenching by addition of 1 N NaOH, the mixture was extracted with AcOEt. The combined extracts were washed with saturated NaCl, dried (Na₂SO₄), and concentrated to afford 3b (3.79 g, 99%) as a colorless oil, which was used for the next reaction without further purification; FTIR (KBr) ν 1705, 1441, 1376, 1244, 1223, 1085 cm⁻¹; ¹³C NMR (100 MHz, CDCl₃) δ 24.5 (×2), 44.9, 52.1, 63.0, 88.2, 106.7, 142.2, 154.3; Anal. Calcd for $C_9H_{15}NO_3$: C, 58.36; H, 8.16; N, 7.56. Found: C, 58.41; H, 8.55; N, 7.43.

tert-Butyl 2,2-dimethyl-5-oxo-1,3-oxazinane-3-carboxylate (4a)

A solution of 3a (1.95 g, 8.58 mmol) in CH_2Cl_2 (120 mL) was stirred with bubbling of gaseous ozone at −78 °C. After stirring for 0.5 h, the excess of ozone was removed by passing through nitrogen for 5 min. Then $Me₂S$ (3 mL) was added, and the mixture was allowed to warm to room temperature. After stirring

for 5 h, the mixture was concentrated under reduced pressure, and the residue was recrystallized from hexane–AcOEt to afford 4a (1.65 g, 84%) as colorless prisms; Mp 110–113 °C (from hexane–AcOEt); FTIR (KBr) *v* 1753, 1699, 1387, 1367 cm⁻¹;
¹H NMR (400 MHz, CDCL) δ 1.47 (9H s) 1.69 (6H s) 4.13 ¹H NMR (400 MHz, CDCl₃) δ 1.47 (9H, s), 1.69 (6H, s), 4.13 (4H, s); ¹³C NMR (100 MHz, CDCl₃) δ 24.3 (×2), 28.3 (×3), 50.3, 67.0, 81.1, 88.5, 153.3, 206.3; Anal. Calcd for C11H19NO4: C, 57.62; H, 8.35; N, 6.11. Found: C, 57.86; H, 8.61; N, 6.17.

Methyl 2,2-dimethyl-5-oxo-1,3-oxazinane-3-carboxylate (4b)

To a solution of 3b (3.79 g, 20 mmol) in 1,4-dioxane–H₂O (100 mL, 3 : 1) were added at room temperature 2,6-lutidine $(4.77 \text{ mL}, 41 \text{ mmol})$, OsO_4 $(2.8\% \text{ in H}_2\text{O}, 3.72 \text{ g}, 0.4 \text{ mmol})$, and NaIO_4 (17.5 g, 82 mmol), and the mixture was stirred for 1 h. After dilution by addition of $H₂O$, the mixture was extracted with $CH₂Cl₂$. The combined extracts were washed with saturated NaCl, dried $(Na₂SO₄)$, and concentrated under reduced pressure. The residue was purified by silica-gel column chromatography (eluted with hexane–AcOEt = $5:1$) to afford 4b (2.9 g, 76%) as a colorless solid; Mp 29–31 °C; FTIR (KBr) ν 1753, 1709, 1446, 1374 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.71 (6H, s), 3.70 (3H, s), 4.15 (2H, s), 4.17 (2H, s); 13C NMR (100 MHz, CDCl3) δ 23.9 (×2), 50.2, 52.6, 66.9, 88.7, 154.3, 205.6; Anal. Calcd for $C_8H_{13}NO_4$: C, 51.33; H, 7.00; N, 7.48. Found: C, 51.53; H, 7.27; N, 7.58.

General procedure for the aldol reaction of 4

To a solution of 4 (0.5 mmol) in i-PrOH (1.25 mL) were added aldehyde (0.25 mmol) and catalyst (0.075 mmol), and the mixture was stirred at room temperature. After completion of the reaction, the mixture was diluted with $H₂O$, and extracted with AcOEt. The combined extracts were washed with saturated NH₄Cl and saturated NaCl, dried (MgSO₄), and concentrated under reduced pressure. The residue was purified by silica-gel column chromatography (hexane–AcOEt) to afford the desired product as summarized in Table 1.

After acetylation $(Ac₂O, pyridine, room temperature, over$ night), the major product was isolated in a pure form and its ee value was determined by chiral HPLC analysis using a Chiralcel OD-H. The results are summarized in Tables 1 and 2.

tert-Butyl (6S,1′S)-6-(acetoxy(4-nitrophenyl)methyl)-2,2 dimethyl-5-oxo-1,3-oxazinane-3-carboxylate (5a-Ac)

Pale yellow oil; $[\alpha]_{D}^{18}$ -38.0 (c 1.64, CHCl₃); FTIR (KBr) ν 1752, 1700, 1525, 1370, 1348, 1230, 1154 cm⁻¹; ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$ δ 1.44 (9H, s), 1.53, 1.73, 2.16 (each 3H, s), 3.40 (1H, dd, $J = 17.6$, 1.0 Hz), 4.39 (1H, d, $J = 17.6$ Hz), 4.63 (1H, dd, $J = 4.0$, 1.0 Hz), 6.25 (1H, d, $J = 4.0$ Hz), 7.55 (2H, d, $J = 8.8$ Hz), 8.17 (2H, d, $J = 8.8$ Hz); ¹³C NMR (100 MHz, CDCl3) δ 20.9, 24.4 (×2), 28.3 (×3), 50.4, 72.4, 76.2, 81.4, 89.4, 123.1 (×2), 129.2 (×2), 142.4, 147.9, 153.0, 169.5, 203.7; HRMS Calcd for $C_{20}H_{26}N_2O_8$ 422.1689, Found 422.1684. The ee was determined by chiral HPLC (hexane–i-PrOH = $90:10$, 0.5 mL min⁻¹): t_R 14.57 (minor) and t_R 20.79 (major) min.

Methyl (6S,1′S)-6-(acetoxy(4-nitrophenyl)methyl)-2,2-dimethyl-5-oxo-1,3-oxazinane-3-carboxylate (5b-Ac)

Pale yellow oil; $[\alpha]_{D}^{18}$ -37.7 (c 0.82, CHCl₃); FTIR (KBr) ν 1752, 1709, 1524, 1373, 1348, 1230 cm−¹ ; 1 H NMR (500 MHz, CDCl₃) δ 1.55, 1.76, 2.15 (each 3H, s), 3.46 (1H, dd, $J = 17.5$, 1.0 Hz), 3.67 (3H, s), 4.40 (1H, br d, $J = 17.5$ Hz), 4.64 (1H, dd, $J = 4.0, 1.0$ Hz), 6.24 (1H, d, $J = 4.0$ Hz), 7.54 (2H, d, $J = 9.0$ Hz), 8.17 (2H, d, $J = 9.0$ Hz); ¹³C NMR (125.8 MHz, CDCl₃) δ 20.9, 23.8, 24.3, 50.4, 52.7, 72.4, 76.2, 89.6, 123.1 (×2), 129.2 (×2), 142.2, 147.8, 154.1, 169.5, 203.1; HRMS Calcd for $C_{17}H_{20}N_2O_8 + H$ 381.1298, Found 381.1288. The ee was determined by chiral HPLC (hexane– i -PrOH = $90:10$, 0.5 mL min⁻¹): t_R 31.65 (minor) and t_R 44.90 (major) min.

Methyl (6S,1′S)-6-(acetoxy(2,4-dinitrophenyl)methyl)-2,2 dimethyl-5-oxo-1,3-oxazinane-3-carboxylate (8-Ac)

Pale yellow oil; $[\alpha]_{D}^{24}$ –25.2 (c 0.30, CHCl₃); FTIR (KBr) ν 1755, 1709, 1607, 1539, 1446, 1372, 1350, 1226 cm⁻¹; ¹H NMR (500 MHz, CDCl3) δ 1.58, 1.69, 2.19 (each 3H, s), 3.68 (3H, s), 3.87 (1H, br d, $J = 17.0$ Hz), 4.49 (1H, br d, $J = 17.0$ Hz), 4.56 (1H, br d, $J = 4.5$ Hz), 6.74 (1H, d, $J = 4.5$ Hz), 7.92 $(1H, d, J = 8.5 Hz)$, 8.47 (1H, dd, $J = 8.5$, 2.5 Hz), 8.88 (1H, d, $J = 2.5$ Hz); ¹³C NMR (125.8 MHz, CDCl₃) δ 20.7, 23.7, 24.4, 50.7, 52.8, 69.8, 75.1, 90.1, 120.2, 127.2, 130.5, 138.6, 147.5, 147.9, 169.0, 202.9; HRMS Calcd for $C_{17}H_{19}N_3O_{10}$ + H 426.1149, Found 426.1137. The ee was determined by chiral HPLC (hexane–i-PrOH = $80:20$, 0.5 mL min⁻¹): t_R 38.45 (minor) and t_{R} 63.25 (major) min. Murby (6S1'59-éneressy6-tatrophenybmehyb22-dimenthy-

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Methyl (6S,1′S)-6-(acetoxy(4-trifluoromethylphenyl)methyl)-2,2 dimethyl-5-oxo-1,3-oxazinane-3-carboxylate (9-Ac)

Colorless oil; $[\alpha]_D^{18}$ –45.2 (c 1.11, CHCl₃); FTIR (KBr) v 1752, 1712, 1374, 1327, 1231, 1169, 1125 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.56, 1.74, 2.13 (each 3H, s), 3.46 (1H, dd, $J = 17.1$, 1.5 Hz), 3.66 (3H, s), 4.37 (1H, br d, $J = 17.1$ Hz), 4.63 (1H, dd, $J = 4.2, 1.5$ Hz), 6.21 (1H, d, $J = 4.1$ Hz), 7.50 (2H, d, $J_{AB} = 8.3$ Hz), 7.58 (2H, d, J_{AB} = 8.3 Hz); ¹³C NMR (100 MHz, CDCl₃) δ 20.9, 23.7, 24.4, 50.5, 52.6, 72.8, 76.3, 89.5, 119.9, 122.6, 124.8 (5), 124.8(9), 124.9(3), 124.9(7), 125.3, 128.0, 128.6 (×2), 130.1, 130.4, 130.7, 131.0, 139.1, 154.1, 169.5, 203.2; HRMS Calcd for $C_{18}H_{20}F_3NO_6 + H_404.1321$, Found 404.1321. The ee was determined by chiral HPLC (hexane–i-PrOH = $95:5$, 0.5 mL min⁻¹): t_R 20.35 (minor) and t_R 28.52 (major) min.

Methyl (6S,1′S)-6-(acetoxy(4-cyanophenyl)methyl)-2,2-dimethyl-5-oxo-1,3-oxazinane-3- carboxylate (10-Ac)

Colorless oil; $[\alpha]_D^{18}$ –34.5 (c 1.04, CHCl₃); FTIR (KBr) v 2229, 1750, 1709, 1373, 1231 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.55, 1.74, 2.14 (each 3H, s), 3.45 (1H, dd, $J = 17.2$, 2.0 Hz), 3.67 (3H, s), 4.39 (1H, br d, $J = 17.2$ Hz), 4.61 (1H, br d, $J =$ 4.0 Hz), 6.19 (1H, d, $J = 4.0$ Hz), 7.48 (2H, d, $J_{AB} = 7.6$ Hz), 7.62 (2H, d, J_{AB} = 7.6 Hz); ¹³C NMR (100 MHz, CDCl₃) δ 20.9, 23.8, 24.3, 50.4, 52.7, 72.6, 76.2, 89.6, 112.4, 118.5, 129.0 (×2), 131.8 (×2), 140.3, 154.2, 169.5, 203.2; HRMS Calcd for

Methyl (6S,1′S)-6-(acetoxy(5-nitro-2-furyl)methyl)-2,2-dimethyl-5-oxo-1,3-oxazinane-3-carboxylate (11-Ac)

Yellow oil; $[\alpha]_D^{23}$ –20.4 (c 0.96, CHCl₃); FTIR (KBr) v 1758, 1709, 1537, 1504, 1372, 1358, 1226 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 1.58, 1.80, 2.16 (each 3H, s), 3.70 (3H, s), 3.83 (1H, d, $J = 17.5$ Hz), 4.55 (1H, br d, $J = 17.5$ Hz), 4.74 (1H, d, $J =$ 5.0 Hz), 6.24 (1H, d, $J = 5.0$ Hz), 6.61 (1H, d, $J = 3.5$ Hz), 7.25 (1H, d, $J = 3.5$ Hz); ¹³C NMR (125.8 MHz, CDCl₃) δ 20.7, 23.7, 24.3, 50.4, 52.8, 66.4, 74.1, 90.1, 111.9, 113.0, 151.8 (×2), 157.7, 169.2, 202.2; HRMS Calcd for $C_{15}H_{18}N_2O_9$ + H 371.1091, Found 371.1084. The ee was determined by chiral HPLC (hexane–i-PrOH = $80:20$, 0.5 mL min⁻¹): t_R 38.45 (minor) and t_R 63.25 (major) min.

Methyl (6S,1′S)-6-(acetoxy(4-bromophenyl)methyl)-2,2 dimethyl-5-oxo-1,3-oxazinane-3-carboxylate (12-Ac)

Colorless oil; $[\alpha]_D^{17}$ –28.4 (c 1.21, CHCl₃); FTIR (KBr) v 1751, 1710, 1373, 1232 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.56, 1.73, 2.11 (each 3H, s), 3.45 (1H, dd, $J = 17.1$, 1.2 Hz), 3.66 $(3H, s)$, 4.34 (1H, br d, $J = 17.1$ Hz), 4.59 (1H, dd, $J = 4.2$, 1.2 Hz), 6.11 (1H, d, $J = 4.2$ Hz), 7.25 (2H, m), 7.44 (2H, m); ¹³C NMR (125.8 MHz, CDCl₃) δ 21.2, 23.9, 24.7, 50.6, 52.8, 73.0, 76.5, 89.6, 122.8, 130.2 (×2), 131.3 (×2), 134.3, 154.3, 169.8, 203.6; HRMS Calcd for $C_{17}H_{20}BrNO_6 + H$ 414.0552, Found 414.0531. The ee was determined by chiral HPLC (hexane– i -PrOH = 95 : 5, 0.5 mL min⁻¹): t_R 21.99 (minor) and t_R 28.85 (major) min.

Methyl (6S,1′S)-6-(acetoxy(4-chlorophenyl)methyl)-2,2 dimethyl-5-oxo-1,3-oxazinane-3-carboxylate (13-Ac)

Colorless oil; $[\alpha]_D^{17}$ –37.3 (c 0.77, CHCl₃); FTIR (KBr) v 1750, 1710, 1373, 1232 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 1.56, 1.73, 2.12 (each 3H, s), 3.44 (1H, dd, $J = 17.0$, 1.5 Hz), 3.66 (3H, s), 4.34 (1H, br d, $J = 17.0$ Hz), 4.58 (1H, br d, $J = 4.0$ Hz), 6.12 (1H, d, $J = 4.0$ Hz), 7.26–7.31 (4H, m); ¹³C NMR (100 MHz, CDCl3) δ 21.3, 24.0, 24.8, 50.8, 52.9, 73.1, 76.7, 89.7, 128.5 (×2), 130.0 (×2), 134.0, 134.7, 154.4, 169.9, 203.7; HRMS Calcd for $C_{17}H_{20}CINO_6 + H 370.1057$, Found 370.1037. The ee was determined by chiral HPLC (hexane–*i*-PrOH = 95 : 5, 0.5 mL min⁻¹): t_R 23.14 (minor) and t_R 28.06 (major) min.

Methyl (6S,1′S)-6-(acetoxy(phenyl)methyl)-2,2-dimethyl-5-oxo-1,3-oxazinane-3-carboxylate (14-Ac)

Colorless oil; $[\alpha]_D^{18}$ –88.8 (c 0.66, CHCl₃); FTIR (KBr) v 1751, 1709, 1373, 1233 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 1.56, 1.71, 2.12 (each 3H, s), 3.44 (1H, dd, $J = 17.0$, 1.0 Hz), 3.65 $(3H, s)$, 4.31 (1H, br d, $J = 17.0$ Hz), 4.60 (1H, dd, $J = 4.0$, 1.0 Hz), 6.14 (1H, d, $J = 4.0$ Hz), 7.29–7.31 (3H, m), 7.35–7.37 (2H, m); ¹³C NMR (125.8 MHz, CDCl₃) δ 21.1, 23.7, 24.6,

50.5, 52.6, 73.6, 76.6, 89.3, 128.0 (×2), 128.2 (×2), 128.5, 135.1, 169.7, 203.5 (carbamate carbon is missing); HRMS Calcd for $C_{17}H_{21}NO_6 + H_336.1447$, Found 336.1447. The ee was determined by chiral HPLC (hexane–i-PrOH = $95:5$, 0.5 mL min⁻¹): t_R 24.94 (minor) and t_R 35.08 (major) min.

Methyl (6S,1′S)-(acetoxy(naphthalen-2-yl)methyl)-2,2-dimethyl-5-oxo-1,3-oxazinane-3-carboxylate (15-Ac)

Colorless oil; $[\alpha]_D^{18}$ –16.2 (c 0.98, CHCl₃); FTIR (KBr) v 1750, 1709, 1372, 1231 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.56, 1.74, 2.14 (each 3H, s), 3.42 (1H, dd, $J = 16.8$, 1.4 Hz), 3.63 $(3H, s)$, 4.31 (1H, br d, $J = 16.8$ Hz), 4.70 (1H, dd, $J = 4.2$, 1.2 Hz), 6.33 (1H, d, $J = 4.2$ Hz), 7.45–7.50 (2H, m), 7.53 (1H, dd, $J = 8.5, 1.7$ Hz), 7.78–7.84 (4H, m); ¹³C NMR (100 MHz, CDCl3) δ 21.0, 23.7, 24.5, 50.5, 52.5, 73.6, 76.7, 89.3, 125.7, 126.1, 126.3, 127.5, 127.6, 127.8, 128.2, 132.6, 132.7, 133.2, 154.1, 169.7, 203.5; HRMS Calcd for $C_{21}H_{23}NO_6$ + H 386.1604, Found 386.1597. The ee was determined by chiral HPLC (hexane–*i*-PrOH = $95:5$, 0.5 mL min⁻¹): t_R 27.35 (minor) and t_R 33.88 (major) min.

Derivatization of 5b to the corresponding dibenzoate esters 6 and 7

To a solution of 5b (32 mg, 0.094 mmol) in THF–MeOH (1.8 mL, 5:1) was added at -10 °C NaBH₄ (6.8 mg, 0.178 mmol), and the mixture was stirred at this temperature for 0.5 h. After quenching by addition of ice-water, the aqueous layer was extracted with AcOEt. The combined extracts were washed with saturated NaCl, dried $(Na₂SO₄)$, and concentrated under reduced pressure. The residue was dissolved in $CH₂Cl₂$ (1 mL) , and i -Pr₂NEt (36.4 mg, 0.282 mmol), 4-dimethylaminopyridine (4.6 mg, 0.038 mmol), and p-bromobenzoyl chloride (51.9 mg, 0.235 mmol) were added at room temperature, and the mixture was stirred for 1 h. After quenching by addition of saturated NaHCO₃, the mixture was extracted with AcOEt. The combined extracts were washed with saturated NaCl, dried (MgSO₄), and concentrated under reduced pressure. The residue was purified by silica-gel column chromatography (eluted with hexane–AcOEt = 4 : 1) to afford 6 (21 mg, 31%) as an oil and 7 (10 mg, 15%) as colorless crystals.

Compound 6

Colorless oil; $[\alpha]_D^{24}$ +22.3 (c 1.30, CHCl₃); FTIR (neat) v 1727, 1607, 1590, 1525 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 1.31, 1.54 (each 3H, s), 3.43 (1H, dd, $J = 14.0$, 6.5 Hz), 3.64 (3H, s), 4.33 (1H, dd, $J = 14.0$, 6.5 Hz), 4.35 (1H, dd, $J = 10.0$, 4.0 Hz), 5.61 (1H, dt, $J = 6.5$, 4.0 Hz), 6.10 (1H, d, $J = 10.0$ Hz), 7.50–7.54 (4H, m), 7.63 (2H, d, $J = 8.5$ Hz), 7.67 (2H, d, $J =$ 8.0 Hz), 7.74 (2H, d, $J = 8.5$ Hz), 8.20–8.22 (2H, m); ¹³C NMR (125.8 MHz, CDCl3) δ 22.8, 25.0, 42.8, 52.5, 68.0, 71.9, 72.5, 89.7, 123.5 (×2), 127.5, 127.8, 128.1 (×2), 128.7, 128.9, 130.9 (×2), 131.1 (×2), 131.7 (×2), 131.8 (×2), 144.9, 147.8, 154.7, 163.9, 164.7; HRMS Calcd for $C_{29}H_{26}Br_2N_2O_9 + H 705.0083$, Found 705.0074.

Compound 7

Mp 170–172 °C (from Et₂O); $[\alpha]_D^{21}$ –86.0 (c 0.10, CHCl₃); FTIR (KBr) v 1722, 1700, 1590, 1525 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 1.50, 1.71 (each 3H, s), 3.33 (1H, dd, $J = 14.5$, 3.0 Hz), 3.61 (3H, s), 4.14 (1H, d, $J = 14.5$ Hz), 4.27 (1H, t, $J = 6.0$ Hz), 5.35 (1H, m), 6.17 (1H, d, $J = 5.5$ Hz), 7.47 (2H, m), 7.58 (2H, m), 7.67 (2H, m), 7.76 (2H, m), 7.79 (2H, m), 8.20 (2H, m); 13C NMR (125.8 MHz, CDCl₃) δ 23.4, 24.7, 43.3, 52.4, 71.1, 73.2, 75.0, 89.6, 123.5 (×2), 127.8, 128.0, 128.4 (×2), 128.8, 128.9, 131.0 (×4), 131.8 (×4), 143.2, 147.9, 154.8, 164.3, 164.8; HRMS Calcd for $C_{29}H_{26}Br_2N_2O_9$ + H 705.0083, Found 705.0076.

Mesylation of 5b

To a solution of $5b$ (111 mg, 0.33 mmol) in CH₂Cl₂ (6.7 mL) were added at -10 °C Et₃N (57 mg, 0.56 mmol) and MsCl (56 mg, 0.50 mmol), and the mixture was stirred at this temperature for 0.5 h. After quenching by addition of $H₂O$, the mixture was extracted with CHCl₃. The combined extracts were washed with saturated NH₄Cl, dried (MgSO₄), and concentrated under reduced pressure. The residue was purified by silica-gel column chromatography (eluted with hexane–AcOEt = $1:4$) to afford 16 (117 mg, 86%) as a pale yellow solid; $[\alpha]_D^{21}$ -15.3 (c 0.29, CHCl3); FTIR (KBr) ν 1752, 1708, 1525, 1446, 1369, 1350 cm−¹ ; 1 H NMR (500 MHz, CDCl3) δ 1.56, 1.80, 3.00 (each 3H, s), 3.48 (1H, dd, $J = 18.0$, 1.0 Hz), 3.68 (3H, s), 4.44 (1H, br d, $J = 18.0$ Hz), 4.76 (1H, dd, $J = 4.0$, 1.0 Hz), 6.05 (1H, d, $J = 4.0$ Hz), 7.60 (2H, d, $J = 8.5$ Hz), 8.22 (2H, d, $J =$ 8.5 Hz); ¹³C NMR (125.8 MHz, CDCl₃) δ 23.9, 24.2, 39.1, 50.4, 52.8, 77.2, 78.4, 90.0, 123.5 (×2), 129.3 (×2), 140.4, 148.3, 154.1, 202.3; HRMS Calcd for $C_{16}H_{20}N_2O_9S + H$ 417.0968, Found 417.0954. 30.5, 32.6, 73.6, 76.6, 89.3, 1326 (c3), 1323 (c2), 1325, 2020 Mpound 7

135.1, 169.7, 200.5 (carbumate carbons in minim): 118.86 March 212 (cfom BeO), [*α*β]² -86.0 (c-0.0, CHC), FTR

Could be C_{er}U, NO_U (11 York 1

Reduction of 16

To a solution of 16 (244 mg, 0.585 mmol) in THF–MeOH (6 mL, 5 : 1) was added at −10 °C NaBH4 (27 mg, 0.71 mmol), and the mixture was stirred at this temperature for 0.5 h. After quenching by addition of ice-water, the mixture was extracted with AcOEt. The combined extracts were washed with saturated NaCl, dried $(Na₂SO₄)$, and concentrated under reduced pressure. The residue was purified by silica-gel column chromatography (eluted with hexane–AcOEt = $1:2$) to afford 17 (208 mg, 85%) as an amorphous solid; $\left[\alpha\right]_D^{21}$ +35.9 (c 0.50, CHCl₃); FTIR (KBr) v 3424, 1695, 1525, 1451, 1389, 1350, 1177 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.54, 1.62 (each 3H, s), 2.59 (1H, d, J = 7.6 Hz), 2.92 (3H, s), 3.43 (1H, br t, $J = 8.8$ Hz), 3.69 (4H, br s), 3.89 (1H, m), 4.44 (1H, ddd, $J = 9.4$, 6.8, 2.8 Hz), 5.56 (1H, d, $J = 8.8$ Hz), 7.65, 8.29 (each 2H, d, $J = 8.8$ Hz); ¹³C NMR (100 MHz, CDCl3) δ 24.3, 26.0, 39.0, 46.4, 52.3, 71.9, 77.2, 80.6, 94.7, 123.9 (×2), 128.5 (×2), 143.0, 148.3, 152.8; HRMS Calcd for $C_{16}H_{21}N_2O_9S$ + H 419.1124, Found 419.1133.

Conversion of 17 to aza-sugar 18

To a solution of 17 (50 mg, 0.12 mmol) in MeOH (5 mL) was added at room temperature HCl in MeOH (0.5 mL), and the mixture was stirred for 12 h. After completion of the reaction, the solvent was removed in vacuo, and the residue was diluted with THF (1 mL) and then DBU (2 drops) was added at room temperature. After stirring for 10 min, the mixture was concentrated under reduced pressure and purified by silica-gel column chromatography (eluted with AcOEt) to afford 18 (28 mg, 82%) as colorless crystals; mp 105–108 °C; $[\alpha]_D^{24}$ +48.0 (c 1.09, MeOH); FTIR (KBr) ν 3322, 1687, 1604, 1549, 1518, 1351 cm⁻¹; ¹H NMR (500 MHz, CD₃OD, H-H and C-H COSY, NOESY) δ 2.98 (1H, br d, J = 4.5 Hz; NCH(Ar)CHOH), 3.15 (1H, dd, $J = 13.5$, 6.5 Hz; CHHN), 3.27 (1H, dd, $J = 13.5$, 6.5 Hz; CHH'N), 3.49 (3H, s; NCO₂CH₃), 3.58 (1H, dt, $J = 6.5$, 4.5 Hz; NCH2CHOH), 3.89 (1H, br s; NCH(Ar)CHOH), 7.40 (2H, d, $J = 8.0$ Hz; 2-CH₂ of C₆H₄-4-NO₂), 8.10 (2H, d, $J = 8.0$ Hz; 3-C H_2 of C₆H₄-4-NO₂). ¹³C NMR (125.8 MHz, CD₃OD, C–H COSY) δ 45.0 (CH₂N), 52.6 (NCO₂CH₃), 55.8 (NCHAr), 65.7 (NCH(Ar)CHOH), 71.3 (NCH₂CHOH), 124.6 (\times 2) (aromatic CH), 127.7 (\times 2) (aromatic CH), 146.4 (quaternary aromatic C), 149.1 (quaternary aromatic C), 159.6 (NCO₂CH₃); HRMS Calcd for $C_{12}H_{15}N_2O_6$ + H 283.0903, Found 283.0933.

Typical procedure for the Mannich reaction

To a solution of p-nitrobenzaldehyde (43.8 mg, 0.29 mmol) and p-anisidine (35.8 mg, 0.29 mmol) in DMF (1 mL) were added the catalyst $(12.2 \text{ mg}, 0.09 \text{ mmol})$ and $4a$ $(100 \text{ mg},$ 0.436 mmol), and the mixture was stirred at room temperature for 96 h. Then H_2O and AcOEt were added with vigorous stirring, and the aqueous layer was extracted with AcOEt. The combined extracts were washed with saturated NH4Cl and saturated NaCl, dried $(Na₂SO₄)$, and concentrated under reduced pressure. The residue was purified by silica-gel column chromatography (eluted with hexane–AcOEt–CHCl₃ = $4:1:1$) to afford 19 (88 mg, 63%; syn : anti = 84 : 16) as an inseparable mixture of diastereomers; amorphous solid; FTIR (KBr) ν 1749, 1698, 1605, 1515, 1369, 1347, 1243, 1153 cm−¹ .

Syn-adduct (major product)

¹H NMR (400 MHz, CDCl₃) δ 1.45 (9H, s), 1.54 (3H, s), 1.67 $(3H, s)$, 3.67 $(3H, s)$, 3.71 $(1H, dd, J = 16.8, 1.2 Hz)$, 4.39 $(1H,$ br s), 4.48 (1H, d, $J = 16.8$ Hz), 5.11 (1H, d, $J = 2.2$ Hz), 6.53 (2H, m), 6.68 (2H, m), 7.52(4) (2H, d, $J = 8.8$ Hz), 8.16 (2H, m); ¹³C NMR (100 MHz, CDCl₃) δ 24.3, 24.8, 28.2(3) (\times 3), 50.7, 55.5, 57.4, 78.3, 81.4, 89.2(9), 114.8 (×2), 115.6 (×2), 123.6 (×2), 128.1 (×2), 139.2, 145.5, 147.4(3), 147.3(3), 152.8 (6), 204.9.

Anti-adduct (minor product)

¹H NMR (400 MHz, CDCl₃) δ 1.43 (9H, s), 1.61 (3H, s), 1.74 $(3H, s)$, 3.19 (1H, dd, $J = 16.8$, 1.2 Hz), 3.69 (3H, s), 4.31 (1H, d, $J = 16.8$ Hz), 4.65 (1H, d, $J = 3.9$ Hz), 5.00 (1H, d, $J = 3.9$ Hz), 6.57 (2H, m), 6.70 (2H, m), 7.52(1) (2H, d, $J = 8.8$ Hz), 8.11 (2H, m); ¹³C NMR (100 MHz, CDCl₃) δ 24.3, 24.8, 28.2 (0) (×3), 50.4, 55.5, 57.7, 77.5, 81.4, 89.3(3), 114.8 (×2), 116.0 $(x2)$, 123.1 $(x2)$, 129.6 $(x2)$, 139.1, 145.5, 147.3(9), 147.4(3), 152.9(3), 205.4.

Derivatization of 19 to the corresponding mono-acetate

To a solution of 19 (310 mg, 0.64 mmol) in MeCN (5 mL) was added 1 N HCl (15 drops) at room temperature, and the mixture was stirred for 24 h. After dilution by addition of CH_2Cl_2 , the mixture was dried (K_2CO_3) and concentrated. The residue was dissolved in CH_2Cl_2 (5 mL) and Ac₂O (0.25 mL) and pyridine (0.5 mL) were added at room temperature. After stirring for 9 h at room temperature, the reaction was quenched by addition of H2O and extracted with AcOEt. The combined extracts were washed with 1 N HCl, saturated NaHCO₃, and saturated NaCl, dried (MgSO4), and concentrated under reduced pressure. The residue was purified by silica-gel column chromatography (hexane–acetone–CHCl₃) to afford the mono-acetate $19-Ac$ (176 mg, 57%); yellowish oil; $[\alpha]_D^{24}$ +14.4 (c 0.175, CHCl₃); FTIR (neat) v 3381, 1739, 1708, 1515, 1367, 1347, 1231 cm⁻¹;
¹H NMR (500 MHz, CDCL) δ 1.45 (9H s), 2.11 (3H s), 3.68 ¹H NMR (500 MHz, CDCl₃) δ 1.45 (9H, s), 2.11 (3H, s), 3.68 $(3H, s)$, 4.05 (1H, dd, $J = 20.0, 5.0$ Hz), 4.25 (1H, dd, $J = 20.0$, 5.0 Hz), 5.16 (1H, br s), 5.19 (1H, m), 5.52 (1H, d, $J = 3.0$ Hz), 6.50 (2H, m), 6.69 (2H, m), 7.50 (2H, d, $J = 8.5$ Hz), 8.18 (2H, d, $J = 8.5$ Hz); ¹³C NMR (125.8 MHz, CDCl₃) δ 20.4, 28.3 $(\times 3)$, 48.7, 55.6, 58.4, 79.1, 80.4, 114.9 $(\times 2)$, 115.4 $(\times 2)$, 124.0 (×2), 127.9 (×2), 138.9, 145.8, 147.6, 153.1, 155.5, 169.4, 201.4; HRMS Calcd for $C_{24}H_{28}N_3O_8 + H$ 487.1955, Found 487.1947; The ee was determined by chiral HPLC using a Chiralpak AD (hexane–*i*-PrOH = 90 : 10, 0.8 mL min⁻¹): t_R 39.42 (minor) and t_R 44.62 (major) min. nistine use sined for 12 h. After completion of the maxim, **Derivatization of 19 o the corresponding mone-scetare

with THF (1 m) and then DIU (2 dongs was alled at mome To a solution of 19 (210 mg, 6.64 mos) in MCS (5 mL**

Acknowledgements

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Notes and references

- 1 For reviews, see: (a) D. Enders, M. Voith and A. Lenzen, Angew. Chem., Int. Ed., 2005, 44, 1304; (b) M. Markert and R. Mahrwald, Chem.–Eur. J., 2008, 14, 40; (c) D. Enders and A. A. Narine, J. Org. Chem., 2008, 73, 7857.
- 2 Review: S. Knapp, Chem. Rev., 1995, 95, 1859.
- 3 Review: S. M. Lait, D. A. Rankic and B. A. Keay, Chem. Rev., 2007, 107, 767.
- 4 (a) For a recent report, see: V. Jha, N. B. Kondekar and P. Kumar, Org. Lett., 2010, 12, 2762, and references cited therein. See also: (b) F. W. Lewis, M. C. Eichler and D. H. Grayson, Synlett, 2009, 1923; (c) W. S. McCall and D. L. Comins, Org. Lett., 2009, 11, 2940; (d) S. Crotti, F. Bertolini, F. Macchia and M. Pineschi, Org. Lett., 2009, 11, 3762; (e) R. W. Bates and Y. Lu, J. Org. Chem., 2009, 74, 9460; (f) G. T. Rice and C. White, J. Am. Chem. Soc., 2009, 131, 11707; (g) R. Obinata, T. Kawasaki-Takasuka and T. Yamazaki, Org. Lett., 2010, 12, 4316; (h) R. Millet, A. M. Träff, M. L. Petrus and J.-E. Bäckvall, J. Am. Chem. Soc., 2010, 132, 15182; (i) L. A. Boralsky, D. Marston, R. D. Grigg, J. C. Hershberger and J. M. Schomaker, Org. Lett., 2011, 13, 1924.
- 5 For reviews, see: (a) P. I. Dalko and L. Moisan, Angew. Chem., Int. Ed., 2001, 40, 3726; (b) P. I. Dalko and L. Moisan, Angew. Chem., Int. Ed., 2004, 43, 5138; (c) Acc. Chem. Res., 2004, 37, (Issue (8)), ed. K. N. Houk and B. List; (d) Adv. Synth. Catal., 2004, 346, (Issues $(9-10)$), ed. B. List and C. Bolm; (e) Asymmetric Organocatalysis, ed. A. Berkessel and H. Gröger, Wiley-VCH, Weinheim, 2005; (f) H. Pellissier, Tetrahedron, 2007, 63, 9267; (g) H. Kotsuki, H. Ikishima and A. Okuyama, Heterocycles, 2008, 75, 493 & 757 (h) P. Melchiorre, M. Marigo, A. Carlone and G. Bartoli, Angew. Chem., Int. Ed., 2008, 47, 6138; (i) A. Moyano and R. Rios, Chem. Rev., 2011, 111, 4703.
- 6 Review: H. Kotsuki, Y. Komatsu, H. Ikishima, A. Okuyama and M. Nakamura, J. Synth. Org. Chem. Jpn., 2009, 67, 65.
- 7 (a) H.-S. Byun, K. C. Reddy and R. Bittman, Tetrahedron Lett., 1994, 35, 1371; (b) K. Ramarajan, K. Ramalingam, D. J. O'Donnel and K. D. Berlin, Org. Synth. Coll. Vol. VII, 1990, 210; (c) J. Villieras and M. Rambaud, Org. Synth. Coll. Vol. VIII, 1993, 265.
- 8 (a) T. Kano, J. Takai, O. Tokuda and K. Maruoka, Angew. Chem., Int. Ed., 2005, 44, 3055; (b) D. Seebach, A. K. Beck, D. M. Badine, M. Limbach, A. Eschenmoser, A. M. Treasurywala, R. Hobi, W. Prikoszovich and B. Linder, *Helv. Chim. Acta*, 2007, 90, 425; (c) C. Isart, J. Burés and J. Vilarrasa, Tetrahedron Lett., 2008, 49, 5414; (d) X. Companyó, G. Valero, L. Crovetto, A. Moyano and R. Rios, Chem.–Eur. J., 2009, 15, 6564.
- 9 (a) H. Torii, M. Nakadai, K. Ishihara, S. Saito and H. Yamamoto, Angew. Chem., Int. Ed., 2004, 43, 1983; (b) A. J. A. Cobb, D. M. Shaw and S. V. Ley, Synlett, 2004, 558; (c) A. J. A. Cobb, D. M. Shaw, D. A. Longbottom, J. B. Gold and S. V. Ley, Org. Biomol. Chem., 2005, 3, 84; (d) V. Franckevicius, K. R. Knudsen, M. Ladlow, D. A. Longbottom and S. V. Ley, Synlett, 2006, 889.
- 10 H. Ikishima, Y. Sekiguchi, Y. Ichikawa and H. Kotsuki, Tetrahedron, 2006, 62, 311
- 11 CCDC 844613 contains the supplementary crystallographic data‡ for this paper.
- 12 For example, see: (a) J. T. Suri, D. B. Ramachary and C. F. Barbas III, Org. Lett., 2005, 7, 1383; (b) J. T. Suri, S. Mitsumori, K. Albertshofer, F. Tanaka and C. F. Barbas III, J. Org. Chem., 2006, 71, 3822;

(d) R. Dodda and C. G. Zhao, Org. Lett., 2006, 8, 4911; (d) S. S. V. Ramasastry, K. Albertshofer, N. Utsumi, F. Tanaka and C. F. Barbas III, Angew. Chem., Int. Ed., 2007, 46, 5572.

- 13 The absolute configurations of unknown products were surmised by analogy.
- 14 MeCN was used as the solvent for the reaction with 5-nitro-2-furaldehyde due to the low yield $(39\%, syn: anti = 48:52)$ under the standardized conditions (Table 2, entry 4).
- 15 For example, see: (a) B. G. Davis, M. A. T. Maughan, T. M. Chapman, R. Villard and S. Courtney, Org. Lett., 2002, 4, 103; (b) T. M. Chapman, S. Courtney, P. Hay and B. G. Davis, Chem.– Eur. J., 2003, 9, 3397; (c) X. Zhou, W.-J. Liu, J.-L. Ye and P.- Q. Huang, Tetrahedron, 2007, 63, 6346; (d) E.-L. Tsou, S.-Y. Chen, M.-H. Yang, S.-C. Wang, T.-R. R. Cheng and W.-C. Cheng, Bioorg. Med. Chem., 2008, 16, 10198; (e) A. López-Pérez, J. Adrio and J. C. Carretero, J. Am. Chem. Soc., 2008, 130, 10084; (f) J.-B. Behr, Tetrahedron Lett., 2009, 50, 4498; (g) J.-K. Su, Y.-M. Jia, R. He, P.- X. Rui, N. Han, X. He, J. Xiang, X. Chen, J. Zhu and C.-Y. Yu, Synlett, 2010, 1609; (h) A. Mordini, M. Valacchi, F. Epiroti, G. Reginato, S. Cicchi and A. Goti, Synlett, 2011, 235; (i) A. Kotland, F. Accadbled, K. Robeyns and J.-B. Behr, J. Org. Chem., 2011, 76, 4094. I (a) H. S. Posts, X. C. Reddy ard R. Homma, Totalogous Hat, 1994, (d) R. Dadda ard C., G. Zhao, Org. Let, 2006, R. 4911. At Albany on 24 March 2012 Published on 24 March 2012 of Albany 2012 Published on the C. Sharp 4.
	- 16 For example, see: D. Enders, C. Grondal and M. Vrettou, Synthesis, 2006, 3597.